

Mansoura University	Final Exam  Engineering Material II	2 <sup>nd</sup> Year Production Students
Faculty of Engineering		Time: 3 Hours
Prod. & Mech. Design Dept.		June, 2013

Answer All Questions in the same order. Use equations and diagrams whenever possible

- 1 – Choose four different unit cells and calculate the packing factor for each of them .
- 2 – A cubic volume of MgO which is 4.2 angstroms along each edge contains 4 Mg<sup>+2</sup> ions and 4 O<sup>-2</sup> ions . What is the density of MgO ?  
(Mg. at.wt. = 24 amu and at.wt. of O<sub>2</sub>=16 amu) (  $AV. N^{\circ} = 6.02 \times 10^{23} \text{ atom/atomic wt}$  )
- 3 - Discuss Bragg's equation and Debye & Scherrer method .
- 4 – Titanium is BCC in its high-temperature form . The radius increases 2%. when the BCC changes to HCP during cooling . what is the percentage volume change (basis one atom)
- 5 – Draw the five possible binary diagrams .
- 6 – How many grains are observed in a microscope per square inch at magnification (X) 100 linear for grain size number 8 & 5
- 7 – Draw the iron – carbon Equilibrium Diagram .
- 8 – Draw the cooling curves for 1010 , 1020 , 1040 , 1080 and 1090 steels .
- 9 – Calculate the percentages of Ferrite , Cementite and Pearlite for the above mentioned steels at room temperature .
- 10 – Draw the continuous cooling transformation diagram (TTT diagram) for 1040 steel showing the effect of the cooling rate on the austenite transformation .
- 11 – Use the TTT diagram to illustrate the following heat treatment operations :  
A – Annealing B – Normalizing C – Quenching D – Martempering E – Austempering.
- 12 – Discuss and derive Fick's first law of Diffusion .

Good Luck,

Prof. Essayed Abdelrasoul