UTILIZATION EFFICIENCY OF DIFFERENT INDUSTRIAL BYPRODUCTS IN AMELIORATION OF SALINE-SODIC SOILS

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ABSTRACT

A field experiment was conducted during two successive years (٢٠١١/٢٠١٢ and TITY/TIT) in a clay saline soil located at South of El-Hussynia plain, Research Station ,Port-Said Governorate, Egypt. This study aims to evaluate the utilization efficiency of some industrial byproducts i.e. Sugar lime, vinasse, by-pass, pyrite (FeS_Y) and aluminum sulfate (Al_Y (SO_E)_Y \AH_YO) individual and interaction among them as a soil amendments and their effects on improving some chemical and physical properties of saline sodic soils, comparing with gypsum application, under Intermittent leaching technique was adopted using El-Salam canal water. The results showed that the chemical and physical properties of the studied soil were clearly improved due to amendments addition. The common parameters of saline sodic soil i.e., EC, pH, ESP, RSE, B.D, P.R, and WTD were clearly improved. The superior improvement of these parameters was resulted from $T_{\text{\tiny A}},\,T_{\text{\tiny 9}}$ and $T_{\text{\tiny Y}}.$ The amendment can be arranged, ascending as follows: $T_1 < T_2 < T_3 < T_4 < T_5 < T_6 < T_6 < T_1 < T_7 < T_8 < T_9 < T$ Tyr< Tyr< The final aim is to displace Na from an exchange complex and replace it with Ca, leading to a decrease of exchangeable sodium percentage (ESP) or increased removal sodium efficiency (RSE) consequently improvement of soil structure. An additional advantage to this strategy is that any mixture is inexpensive and ready available in large quantities in contrast to gypsum in this study.

Keywords: Saline sodic soils; amelioration, soil amendments, sugar lime, vinasse, By-pass, pyrite, gypsum, alternative gypsum.

INTRODUCTION

Saline-Sodic soils reclamation will be one of the main problems for humans in the future. The reclamation of saline uses many different methods such as physical amelioration (deep ploughing, sub-soiling, sanding and etc.) ,chemical amelioration(amending soil i.e.: gypsum, calcium chloride, sulphuric acid and sulphur,) electro-reclamation (treatment with electric current). The most effective procedures are based on the removal of exchangeable and soluble sodium as well as modification of the ionic composition of soils by adding chemicals paralleled with leaching of sodium salts out of the soil profile (Mahdy, Y+11). The worldwide occurrence of such soils on on x1. har emphasizes the need for efficient, inexpensive, and environmentally acceptable management. These soils can be ameliorated by providing a source of calcium (Ca⁺) to replace excess sodium (Na⁺) from the cation exchange sites. (Shainberg &Letey ۱٩٨٤). The ٦٢٠٠٠ feddans of south EL-Hassanyia plain is situated in the North part of the river Nile Delta. The entire area is devoid of vegetation because of the extremely high salinities. These areas irrigated from EL-Salam canal. This brings water to the area from the river Nile, mixed with water from the Haddous and the lower Sirw

outfall drains in as much as 1: 1,70. Grattan and Oster, (7.17) reported that leaching is the only effective way to decrease excessive salts from the root zone of the salt affected soils. This is the process of dissolving and transporting soluble salts by downward movement of water through the soil surface. Its efficiency can be defined as the quantity of soluble salts leached per unit volume of water applied to the soil. Keren, (1991) showed that intermittent leaching is based on giving a set amount of water to the leaching plot and allowing this set amount to be drained completely to the drains. Sometimes intermittent leaching is combined with mulching to improve its performance. Leaching efficiency increased under intermittent leaching. It allowed more time for the movement of water through pores and improved the leaching efficiency.

Abdalla et al. (Y.Y.) concluded that the drainage installation is the most important tool to conserve or reclaim the harmful effect of salty clavey soils to a feasible one. This process must be under taken with gypsum requirements. The application of gypsum enhances leaching by improving soil hydraulic conductivity (Ghafoor et al. Y ...). The application of gypsum for the reclamation of sodic soils enhanced the removal of soluble Na+, decreased salinity, ESP and pH and increased soluble and exchangeable calcium and hydraulic conductivity of the reclaimed soil. Hussain et al. (Y · · ·) observed that the applications of amendments before leaching improved permeability better than leaching before the application of amendments. The use of sugar lime and vinasse, which are final by-product of the sugar industry, is of great interest because of their low costs and the large quantities that are being produced. Mansour (Y··Y) showed that adding sugar lime to saline sodic soils increased total porosity, water holding capacity, quickly drainable and water holding pores, consequently soil hydraulic conductivity increased. On the other hand, soil bulk density and fine capillary pores were decreased by increasing the application rate. Reda et al. (٢٠٠٦) found that the application of sugar lime with sulphur to saline sodic soil improved soil structure.

Vinasse also is a final by-product of the sugar industry. It is produced after removal of the fermentation products from molasses, it can be characterised by high organic carbon (" $\circ \cdot - \wedge " \circ g \ Kg^-$) and nutrient contents (" $\cdot - \circ " \circ g \ N \ kg^-$) and " $\cdot - \circ " \circ g \ K \ kg^-$) in this by-product make it potentially useful as a fertilizer, although with some constraints to its salinity, low C: N ratio and low phosphorus content. Addition of such by-product as an amendment to soil led to improve the physical, chemical and biological properties of soils, as well as the reduction of disposal costs (Parnaudea *et al.* $^{*} \cdot \cdot ^{*} \wedge ^{*}$).

alumina source and an iron source. Abd El-Hamid et al. (٢٠١١) concluded that the usage of any amendments gypsum, sugar lime, By-pass, mixture (1) and mixture (Y) could be positively affect on about reclamation of saline clay soil in Sahall El-Tina district. Mansour et al.(٢٠١٤) concluded that using a suitable amendments mixtures under suitable application method (surface +subsurface) with intermittent leaching cycles, were the best which led to short time for reclamation of clay saline sodic soils. Dahlya et al. (١٩٨١) observed that leaching intermittently allowed more time for the movement of water through pores and improved the leaching efficiency. Al-Sibai et al. (1997) concluded that intermittent leaching could improve leaching efficiency. Iron pyrites waste products from mining operations and are also mined products, they have been used with varying success to supply Fe and S to plants. Furthermore, ameliorate sodic soils when oxidized to acid. Several forms of pyrite exist depending upon origin and crystallinity.((Wallace &Wallace 1997). Ahmed et.al.(1947) and Mace et al,(1999) found that a comparison of gypsum, pyrite, aluminum sulfate and sulfuric acid in reclamation of sodic soils indicated good possibilities of utilizing pyrite and sulfuric acid for sodic soil amelioration, as well as pyrite can be also used as a source of micronutrients needed for plant growth. Gulsen et al. (Y + 1 2) reported that the application of waste pyrite or sulfuric acid to calcareous sodic soil with dose of ££, V and ro, r Mgha' superior to the gypsum of oo, r Mg ha in terms of ESP in frand fr weeks respectively.

The aim of this experiment was to improve the efficiency of some industrial byproducts i.e. Sugar lime, vinasse, by-pass, pyrite (FeS $_{\tau}$) and aluminum sulfate (AI $_{\tau}$ (SO $_{\tau}$) $_{\tau}$ 1 $_{\tau}$ AH $_{\tau}$ O) individual and interaction among them as soil amendments (alternative gypsum) in saline sodic soils, as well as, to evaluate their effects on improving some soil chemical and physical properties.

MATERIALS AND METHODS

A field experiment was conducted at South-El-Hussynia plain, Agricultural Research Station, Port Said Governorate, during two successive years $(1)^{1/1}$ and $(1)^{1/1}$ to study the utilization efficiency of some industrial by-products materials, as soil amendments on improving the properties of saline sodic soils .Characteristics of the studied experimental soil are presented in Table (1). In addition, the chemical composition of sugar lime, vinasse and By-pass is tabulated in Table (1). The composition and chemical properties of the four mixtures of amendments used are presented in Table (1)

Table 1: Characteristics of the studied experimental soil

Soil properties	Value
Particle size distribution (%)	
Coarse sand	۲,٧
Fine sand	18,7
Silt	77,7
Clay	٦٠,٤
Texture class	Clay
Chemical analysis	
pH (\':\o)	9,17
EC dS.m ⁻	٣٨,٢
ESP(%)	۲۸,۷

Table [†]: Chemical composition of sugar lime, vinasse and by-pass used.

Characteristics	Sugar lime	Vinasse	Cement kiln dust (By-pass)
Density (Mg m ⁻¹)	٠,٧٤	1,15	٠,٦٣
pH (۱:۲,۵)	۸,٣٠	٤,٥٠	17,.
SP	٧٠,٠		۲٠٩
O.M (%)	٣,٤٢	٣٨,٣	-
	Total el	ements (%)	
Nitrogen	٠,٩٤	٠,٢٠	٠,٠٢
Potassium	٠,٠٦	٠,٧١	١,٣٦
Calcium	۲۸,٥	٠,٦٥	٤,٥١
Phosphorus	٠,٢٨	٠,٢١	٠,٠٩
Manganese	٣,٤٢	٠,٦٠	٠,٣٥
Iron	٠,٠٠٧	٠,٠٠٦	٠,٠١١
Copper	٠,٢١	٠,٠٠٧٣	۲,۰۲
Zinc	٠,٠٠٣	٠,٠٠٢٤	٠,٠٠٣

The experimental layout:

The studied soil has a shallow water table ($\mathfrak{t} \cdot \mathsf{cm}$. from the soil surface) which caused lower hydraulic conductivity ($\mathfrak{t},\mathfrak{t} \cdot \mathsf{cmh}^{-1}$) and higher EC = $\mathfrak{T}^{\Lambda},\mathfrak{T}$ dSm $^{-1}$ in saturated paste extract, pH = Λ,\mathfrak{T} in $\mathfrak{T},\mathfrak{o}$ soil water suspension.

The field experimental was tilled by deep plowing (ε cm depth). Calculated the distance between the drainage according to the equation of Hooghoudt ($^{14}\varepsilon$) which was identified at 1 , m. So drainage ditches were drilled on spacing of 1 and 1 m., in the first period (1 months). Where. Intermittent leaching conducted from $^{1/2}(^{1})$ to $^{1}(^{1})$. The leaching water was supplied from irrigation canal lie between two drains, as shown in Fig 1 . The intention in this case was to apply water on surface to a depth of 1 mm and then allows it to infiltrate and evaporate away until the surface become free or standing water. The soil was then left to dry out for some time before the basins were re-irrigated. Intermittent leaching in which pounded water application is interrupted with rest periods allowing redistribution of the salts held in micro pores. At the end of the first period after leaching process EC was decreased to 14 , and ESP to 14 , v

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Table *: Composition and chemical properties of the nine mixtures of amendments used

Mixtures of	Mixtu	res con	npositi	Chemical properties					
amendments	S.L	B.P	٧	Α	AI.	P.	рН	S.D (g L ⁻ ')	Aval.Ca %
M,	-	٤	١	۲	-	-	٧,١١	۲,۸٦	٠,٨
Μ _Υ	٤	-	١	١	-	-	٧,٠٧	٤,٢٤	٢,٤٦
Mτ	٣	۲	١	١	-	-	٧,٠٩	1,58	۲,۹٦
M ¿	-	٤	١	۲	١	-	٦,٩٥	٣,٧٦	١,٦٥
M。	٤	-	١	١	١	-	٦,٨٥	٦,٤٦	٤,٥٨
Mπ	٣	۲	١	١	١	-	٧,٠	۲,۹۸	0,81
M√	-	٤	١	۲	-	١	٧,٠٨	۲,۹٦	1,70
MA	٤	-	١	١	-	١	٧,٠٢	٤,٤٨	٣,٨٤
Мя	٣	۲	١	١	-	١	٧,٠٥	1,70	٣,٧٦

S.L. Sugar lime V: Vinasse B.P. By-pass A: Commercial Sulfuric acid

Al.: Aluminum sulfate P.: Pyrite

S.D: soluble degree Aval.Ca: available calcium

The experimental design:

After the end of the first leaching cycle the experimental design was laid out as a randomized complete block design with 't treatments and "replications. The field experiment was divided into ("t) plots; with plot area of h. m" ('/o" fed). Amendment treatments were applied to the soil on the basis of gypsum requirements (GR), "the Mg fed to reduce ESP to 'o' at depth 'o cm. (FAO, 'the No.), uniformly spread and thoroughly mixed in the soil surface by tillage. Then repeating leaching cycle for three periods. At the end of each leaching process soil samples were taken for chemical analysis The amount of any amendment to be applied for amelioration is based on the amount equivalent to that of gypsum (Table £).

Table 4: Chemical composition and equivalent amount of a chemical amendment that can substitute One Mega (Mg) of gypsum in ameliorating sodic soils.

Amendment	Chemical composition	Amount equivalent to \Mg of gypsum
Gypsum.	CaSO: ۲H+O	1,+
Calcium chloride	CaCl ₇ - ⁷ H ₇ O	٠,٨٥
Calcium carbonate	CaCO _₹	٠,٥٨
Sulfuric acid	H₁SOჴ	٠,٥٧
Ferrous sulfate	FeSO₅ _ [∨] H√O	1,71
Ferric sulfate	Fe ₇ (SO ₅) ₇ _ ⁹ H ₇ O	1,.9
Aluminum sulfate	$AI_{\tau}(SO_{\epsilon})_{\tau}-1$ $^{\Lambda}H_{\tau}O$	1,79
Sulfur	Š	٠,١٩
Pyrite	FeS⊤	٠,٦٣

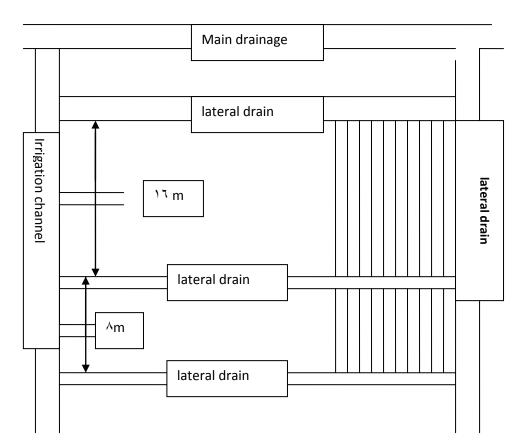


Fig.1: layout of experimental plots in a randomized complete block design

The amount of gypsum needed to ameliorate a specified soil depth is called gypsum requirement (GR).

The treatments of soil amendment materials were:

- Ty Leaching only "not amended"
- T_v Gypsum (G) which added at a rate of \(\cdot\). Mg fed
- T_r Mixture (1) at a rate of r_r Mgfed according to (GR)= 1 · · % GR . T_t Mixture (r) at a rate of r_r Mg fed according to (GR)= 1 · · % GR. T_t Mixture (r) at a rate of r_r Mg fed according to (GR)= 1 · · % GR.
- T₁ Aluminum sulfate which added at a rate of V, Vo Mg fed '.

- Ty Mixture (\$\epsilon\$) at a rate of \$\tau_1^T\$ Mg fed according to (GR) = \frac{1}{2} \% GR T_A Mixture (\$\epsilon\$) at a rate of \$\tau_1^T\$ Mg fed according to (GR) = \frac{1}{2} \% GR T_A Mixture (\$\epsilon\$) at a rate of \$\tau_1^T \cdot \text{Mg fed} according to (GR) = \frac{1}{2} \% GR
- T_1 . Pyrite which added at a rate of r , h Mg fed.
- T₁₁ Mixture ($^{\lor}$) at a rate of $^{\xi}$. Mg fed according to (GR) = $^{\backprime}$. % GR T₁₇ Mixture ($^{\land}$) at a rate of $^{\lnot}$. Mg fed according to (GR) = $^{\backprime}$. % GR
- T_{17} Mixture (9) at a rate of 7,7 Mg fed according to (GR) = 1... % GR

Table o: List of published methods used in the determined properties.

Soil properties	References
Particle size distribution (%)	Gee and Bauder, ۱۹۸٦
Bulk density (Mg m ⁻¹)	Vomocil, 1970
Hydraulic conductivity (cm h ⁻¹)	VanBeers, 1904 (Auger hole method)
Ground water levels	Luthin ١٩٦٦.
Soil pH and electrical conductivity (dS m ⁻¹)	Page et al. 19AY.
Gypsum requirement (Mg fed ^{-'})	FAO, 19AA.

Removal sodium efficiency (RSE):

Removal sodium efficiency is the percentage of Na-removed from soils at end of the experiment was calculated as follows:

RSE = $(ESP_i - ESP_f) / ESP_i \times \cdots$ Where:

ESP_i = exchangeable sodium percentage before the soil amendments application

ESP_f = exchangeable sodium percentage after the soil amendments application at the end of the experiment

RESULTS AND DISCUSSION

Effect of applied soil amendments and leaching process on soil properties:

a.) Soil salinity (EC_e):

The EC_e values of soil samples at the end of every season, in the studied soil under different application rate of amendments and leaching process are presented in Tables (7 and 9) The data indicate that EC_e decreased to 5,1 and 5,4 dSm $^{-1}$ for both section width of 4,1 and 1,7 m. in the surface layer., Also, data showed that there were differences in EC_e under treated soils among depths and active desalinization was observed in all the treatments. The sharp decrease in EC_e in all treatments was observed with T_A and T₃ treatments which were found to be most effective treatments, for section width of 4 .m and 1,7 m. , respectively.

Also, the values of Ec_e can be arranged descending as follows: $T_1 > T_7 > T_7$

enhanced the leaching process. As well as additional open drainage installations and leaching cycle. (Abd El-Hamid, *et al.* ۲۰۱۱).

Data also, showed, that application of any amendment to soils causes a clear decline in the EC_e values compared to EC_e of initial soil (first period). This effect is more pronounced in the surface layer. Surface applied water would pass through the surface applied amendment and infiltrate the top layers allowing exchange process between $Ca^{\tau+}$ and Na^{τ} (El-Sharawy *et al.* $\tau \cdots \wedge$).

b) Soil PH:

Soil pH has a considerable impact on controlling the plant nutrients. particularly the availability of micronutrients (Naidu & Rengasamy 1997). The use of saline-sodic water on soils for agriculture without an amendment application, in general, tends to increase the soil pH that impacts the soil nutrient availability, rendering plants with malnutrition (Curtin & Naidu 1997) In this study, the data in Tables (7 and 7) obtained after four periods revealed differences between the treatments of the soils used .All treatments had the ameliorative potential to decreased the pH value after leaching processes, especially T_A T₃ and T_V, during reclamation of saline-sodic soil. pH of clay saline-sodic soil was less affected than pH of Sandy soil after amendments application because of the high clay content which acted as a buffer and resisted any appreciable change in soil pH in the alkaline range. In general, a high EC to SAR ratio tends to lower pH and vice versa (Ghafoor et al. ۲۰۰۱). Also, the data showed that the using of different forms of soil amendments reduced the pH value. T_A amendment was the most effective in reducing the pH values than other amendments. in both for section width of ^.m and \7 m. at surface layer compared the pH of initial soil. The pH values can be arranged descending as fallows $T_1>T_7>T_1>T_7>T_7>T_9>T_9>T_{17}$ same trend was observed in both for section width of A.m and NT m. for subsurface layers. Similar trend was found throughout the three seasons, this could be attributed to one or more of the following reasons. The presence of soil amendments its constituents (i.e. S.L., B.P., vinasse and sulfuric acid). Vinasse and S.L. can be characterised by high organic carbon (Habib et al. τ···٩). By-pass (B.P.) and sugar lime (S.L.) has high content of CaCO_τ and can be used as a source of Ca^{'+}. Sulfuric acid may increase quantities of exchangeable and soluble calcium. Also presence of Al in Aluminum sulfate, Fe in pyrite (FeS_Y) enhanced the leaching process and the presence of high adsorptive capacity materials like compost adsorb more sodium as well as additional open drainage installations and leaching.

Table " Some chemical properties of soil at surface layer (· - ' o cm.) throughout the experimental time.

throughout the experimental time.										
		Section	Section width of \7 m.							
Treatments	*Periods	Ec _e (dSm ⁻ ')	ESP (%)	RSE (%)	рН	Ec _e (dSm ^{-')}	ESP (%)	RSE (%)	рН	
Intermittent leaching only.	∖ st	19,1	۲۸,۷	-	9,17	19,1	۲۸.۷	-	9,17	
T ₁ Control	۲nd	10,7	۲۷,۳	٤,٨٨	۸,۹۲	۱۷,۸	۲۸,۷	1,.7	۸,9٤	
((Leaching only)	۳ rd	۱۲,۸	۲٥,٤	11,0	۸,۹٦	10,7	۲۸,٥	۲,٦٧	۸,۹۸	
	٤ th	۸,٧	77,1	۲۳,۰	۸,۹۹	11,7	۲۸,۳	٣,٧٤	٩,١١	
Τr	۲ nd	17,7	۲۳,۳	۱۸,۸	۸, ٤٠	15,7	70,7	17,7	۸,٧٠	
	۳ rd	١٠,٦	۲٠,١	٣٠,٠	۸,١٥	11,7	71.1	٤٠,٦	۸,٦٠	
	٤ th	۸,٥	17,7	٤٢,٢	۸,٠٥	٦,٨	۱۸,۳	٣٦,٢	٨,٤٥	
T۴	۲na	17,0	۲۱,۷	۲٤,٤	٧,٩٥	15,1	70,7	۱۸,۷	٨,٥٥	
	۳ rd	١٠,١	19,0	٣٢,١	٧,٨٥	١١,٦	۲۰,٦	٤٣,٣	۸,٤٠	
	٤ th	٧,٨	10,7	٤٧,١	٧,٧٠	٥,٤	17,0	٣٩,٠	۸,٣٠	
Τ٤	۲ nd	١٢,٦	11,0	71,7	٧,٨٥	1 ٤, ٤	75,7	۲۱,٤	٨,٤٥	
	۳ rd	1.,7	19.1	۳۱,۰	٧,٧٥	11,7	71,7	٤٠,١	۸,٣٠	
	٤ th	٧,١	۱۳,۸	01,9	٧,٦٠	٦,٣	١٨,١	٣٦,٩	۸,۲۰	
T。	۲ nd	۱۲,۸	7 £ , ٢	10,7	٧,٩٠	17,7	۲٦,٥	٧,٦٧	٨,٥٠	
	۳ rd	١٠,٧	۲٠,٧	۲٧,٩	٧,٨٠	11,5	27,1	۲۰,٦	۸,٣٥	
	٤ th	٧,٤	۱٤,٨	٤٨,٤	٧,٦٥	٧,٥	۱۸,٦	40,1	۸,۲٥	
Τι	۲ nd	11,7	19,7	٣٢,٨	۸,۰۰	15,7	77,9	۲۰.۲	۸,۱۰	
	۳ rd	٩,٦	۱٦,٥	٤٢,٥	٧,٩٠	۱۱,۸	19,5	٣٢,٨	٨,٤٥	
	٤ th	٧,٩	10,7	٤٥,٦	٧,٧٥	٥,٢	17,7	٤٢,٢	۸,٣٥	
T _Y	۲ nd	۱۱,٤	۱۸,٤	٣٥,٩	٧,٦٥	10,7	۲۱,٦	75,7	۸,۲٦	
	۳ rd	۸,٦	10,7	٤٥,٣	٧,٥,	17,1	۱۸,٦	40,1	۸,۱۰	
	٤ th	٥,٣	11,4	٥٨,٩	٧,٣٥	٩,٢	17,7	٤٣,٦	۸,۰۰	
TΑ	۲ nd	۱٠,٩	19,.	٣٣,٨	٧,٥٥	10,5	71,0	75,7	۸,۱٥	
	۳ rd	٧,٨	۱٦,٠	٤٤,٣	٧,٤٠	11,7	۱۸,۳	٣١,٠	۸,۰۰	
	٤ th	٤,١	11,7	٦٠,٦	٧,٢٥	٤,٨	١٦,٤	٤٢,٢	٧,٩٠	
Τ٩	۲ nd	۱۳,٤	10,1	17,0	٧,٦٠	١٦,٢	۲۷,٥	٨,٥٦	۸,۲۰	
	۳ rd	۱۱,٤	۲۱,۳	Y0,1	٧,٤٥	٩,٢	۲۳,۸	10,1	۸,۰۳	
	٤ th	٤,٦	۱۱,٤	٦٠,٤	٧,٣٠	٧,٩	۱۸,۳	٤٠,٦	٧,٩٢	
Τ,.	۲ nd	۱۲,٤	۲۱,۱	۲٦,٥	۸,٠٥	١٦,٥	۲۳,٦	۱۷,۸	۸,٦٥	
	۳ rd	٩,٩	19,.	٣٣,٨	٧,٩٥	17,7	71,7	۲٦,١	۸,٥١	
	٤ th	۸,۱	۱٦,٤	٤٢,٩	٧,٨٥	٦,٢	۱۷,٦	٣٨,٧	۸,٤٠	
T11	۲ nd	11,7	19,7	٣١,٤	٧,٨٠	10,9	۲۱,۱	۲٦,٥	۸,٤٠	
	۳ rd	٩,٧	١٦,٩	٤١,١	٧,٧٠	۱۲,٤	19,9	٣٠,٧	۸,۲٥	
	٤ th	۸,٦	۱۳,۰	٥٤,٧	٧,٥,	0,0	۱٦,٨	٣٦,٩	۸,۱۷	
Tır	۲ nd	۱۳,۷	۲۰,0	۲۸,٦	٧,٧٠	10,9	۲٣,٤	14,0	۸,٣٠	
	۳ rd	۱۱,۸	۱۸,٤	٣٥,٩	٧,٦٠	17,0	۲٠,٥	۲۸,٦	۸,١٥	
	٤ th	٦,٠	17,1	٥٧,٨	٧,٤٠	٧,٦	١٨,٠	٦١,٠	۸,٠٥	
Tır	۲ nd	1 £ , 1	۲٦,٤	۸,۰۱	٧,٧٣	10,5	۲۷,۳	٤,٨٨	۸,٣٥	
	۳ rd	17,7	77,7	19,7	٧,٦٤	۱۳,۷	۲٤,٧	17,9	۸,۲۰	
	٤ th	٦.٣	17,7	٥٦,١	٧,٤٥	۸,۲	19,7	٣٣,١	۸,۱۰	

RSE: Removal sodium efficiency *Period: \ months Table V: Some chemical properties of soil at sub-surface layer (10 - To

cm.) throughout the experimental time.

Treatments	cm.) throughout the experimental time.											
Intermittent 1st 14,0 74,7 - 9,77 14,0 74,7 - 9,77 18,0 74,7 - 9,77 17,0 74,7 - 9,77 17,0 74,7 - 9,77 17,0 74,7 - 9,77 17,0 74,7 - 9,77 17,0 74,7 - 17,7 74,7							Section width of \\ m.					
Intermittent	Treatments	periods	Ec _e	-	-	рН	Ec _e (dSm ⁻ ')	_	_	рН		
Tr. Control (Leaching only) Trill 15,7 14,7 17,1 14,0 17,4 17,1 17,1 17,1 1,4,5) st		۲۸,۷	-	9,77	19,0	۲۸,۷	-	9,77		
(Leaching only) (## 11,1 14,1 1,1 14,1 1,1 14,1 14,1 14,1	T. O	۲ nd	۱۷,۳	۲۸,٥	١,٠٧	۸,۸۲	11,0	۲۸,٦	۰,٥٣	۸,۹۱		
Tr		,	15,7	۲۸,۲	۲,٦٧	۸,۸٥	١٦.٨	۲۸,٥	١,٠٧	۸,۹۳		
Tr	(Leaching Only)		۱۱,٤	۲۷,۹	٤,٢٨	۸,٩٠	17,0	۲۸,۲	۲,٦٧	۸,۹٧		
Em 1, 7, 1 17, 4 21, 7, 1 17, 4 21, 7, 1 17, 4 21, 7, 1 17, 4 21, 7, 1 17, 6 17, 7 17, 7 17, 7 17, 7 17, 7 17, 7 17, 7 17, 7 17, 7 17, 7 17, 7 17, 7 17, 7 17, 1			۱۲,۸	۲۳,٦	۱۷,۸	۸,٦٥	10,1	۲٦,١	17,9	۸,۷۷		
Tr	T۲	'	11,0		۳۱,۷	۸,٥٦	١٠,٢		٣٨,٠	۸,٧٠		
Tr		-	١٠,٦	17,9	٤١,١	۸,٣٥	٧,٩	15,7	٥٨,٨	۸,٤٨		
To			11,4	۲۳,۰	19,9	۸,٦٠	15,7	40,0	17,1	۸,٧٥		
T:	T۲		٩,٢	19,.	٣٣,٨	۸,٥٠	۱٠,٤	۲۱,۳	٣٩,٦	۸,٦٥		
T:			٨,٤	10,7	٤٥,٦	۸,۳۰	٧,٢	۱٤,٠	٦٢,٦	٨,٤٥		
To												
To	T٤			19,8	۳۲,۸	۸,0٣	11,1		٣٨,٠	۸,٦٥		
To		-										
Eth V,A 10,7 £1,V A,Y0 V,9 1Y,7 1£,V A,£ Ynd 10,0 YY,0 YY,7 A,70 11,7 Y1,0 11,A A,VV Trd 9,V 1V,1 YY,1 A,71 VY,1 Y0,1 E1,0 A,VV												
T ₁ Trd q,v 1v,q rrd q,v rrd q	T.					,			,			
Ti		-				,						
Ty	_					,	· ·		,			
Ty	Ττ								,			
Tv		-	· ·									
th o,Y 15,9 £A,1 A,5Y o,Y 1A,0 £9,Y A,0T Ynd 1.1 Y1,A Y£,9 £A,1 A,2Y 1Y,9 YV,Y A,0Y A,0Y A,0Y A,0Y A,0Y A,0Y A,0	-											
The triangle of triangle o	Ιv	-	· · · · · ·			,						
The triangle of the term of th		-	· · · · · ·			,						
£th £,7 1£,V £A,A A,£ 7,£ 1A,. 01,7 A,00 Tq 10,Y Y£,V 1T,9 A,V0 17,T YA,1 17,. A,A0 Trd 7,£ Y,1 Y,1 A,V 1T,£ Y£,A Y,9 A,A1 £th 0,0 1£,A £A,£ A,EA 1,7 Y,0 £T,4 A,A7 Tnd 11,£ YY,9 Y,Y A,V1 1T,V YA,T Y,1 A,A7 Tnd 11,£ YY,0 Y,Y A,V A,Y Y,7 Y,7 A,A7 Tn Ynd 11,Y YY,T YY,T A,YY YY,Y Y,A Y,A YY,Y A,A7 Tn Ynd 10,Y YY,T YY,T X,YY Y,A YY,A YY,A X,AY Tn Ynd 10,Y YY,T YY,T X,YY YY,A X,AY YY,A X,AY YY,A X,AY	-					,						
$T_{1} = \begin{bmatrix} Y^{nd} & 10,Y & Y\xi,V & 17,9 & A,V0 & 17,T & YA,1 & 17,. & A,A0 \\ Y^{rd} & 7,\xi & Y,1, & Y,. & A,V & 17,\xi & Y\xi,A & Y,9 & A,A1 \\ \xi^{th} & 0, & 1\xi,A & \xiA,\xi & A,\xiA & 1,,Y & Y,0 & \xiT,9 & A,TY \\ Y^{nd} & 11,\xi & YY,9 & Y,Y & A,V1 & 17,V & YA,T & Y,1\xi & A,AV \\ \xi^{th} & 1,V & 1A,A & T\xi,0 & A,V & A,7 & Y\xi,9 & Y,T & A,A1 \\ \xi^{th} & 1,1 & 17, & \xi\xi,T & A,0, & 9,A & Y1, & \xi1,Y & A,70 \\ Y^{nd} & 1,V & YY,T & YY,T & A,VY & 1Y, & YV,V & 0,T0 & A,AT \\ Y^{nd} & 1,V & YY,T & YY,T & A,VY & 1Y, & YV,V & 0,T0 & A,AT \\ \xi^{th} & 7,A & 10,1 & \xiY,\xi & A,\xi0 & Y,A & 1A,A & \xiY,7 & A,7V \\ Y^{nd} & 1,9 & YY,0 & Y1,7 & A,V0 & 17,9 & YV,9 & \xi,YA & A,A0 \\ Y^{nd} & 1,9 & YY,0 & Y1,7 & A,V0 & 17,9 & YV,9 & \xi,YA & A,A0 \\ \xi^{th} & 7,A & 10,1 & \xiY,\xi & A,\xi0 & Y,A & 1A,A & \xiY,7 & A,AV \\ \xi^{th} & 7,0 & Y1,0 & Y1,1 & A,V0 & 17,9 & YV,9 & \xi,YA & A,A0 \\ \xi^{th} & 7,0 & Y,0 & Y1,7 & A,V1 & 9,7 & Y\xi,0 & YY,0 & A,A1 \\ Y^{nd} & 10,0 & Y0,\xi & 11,0 & A,51 & Y,7 & Y,1 & \xi7,1 & A,V1 & A,70 \\ Y^{nd} & 10,0 & Y0,\xi & 11,0 & A,51 & 17,0 & YV,1 & 0,0V & A,V0 \\ T_{17} & Y^{nd} & V,0 & Y,7 & Y9,7 & A,01 & 17,\xi & Y7,7 & A,V1 & A,70 \\ Y^{nd} & 10,0 & Y0,\xi & 11,0 & A,51 & 17,0 & YV,1 & 0,0V & A,V0 \\ Y^{nd} & 10,0 & Y0,\xi & 11,0 & A,51 & 17,0 & YV,1 & 0,0V & A,V0 \\ Y^{nd} & Y,0 & Y,7 & Y9,7 & A,01 & 17,\xi & Y7,7 & A,V1 & A,70 \\ Y^{nd} & Y,0 & Y,7 & Y9,7 & A,01 & 17,\xi & Y7,7 & A,V1 & A,70 \\ Y^{nd} & Y,0 & Y,7 & Y9,7 & A,01 & 17,\xi & Y7,7 & A,V1 & A,70 \\ Y^{nd} & Y,0 & Y,7 & Y9,7 & A,01 & 17,\xi & Y7,7 & A,V1 & A,70 \\ Y^{nd} & Y,0 & Y,7 & Y9,7 & A,01 & 17,\xi & Y7,7 & A,V1 & A,70 \\ Y^{nd} & Y,0 & Y,0 & Y,7 & Y9,7 & A,01 & 17,\xi & Y7,7 & A,V1 & A,70 \\ Y^{nd} & Y,0 & Y,0 & Y,7 & Y9,7 & A,01 & 17,\xi & Y7,7 & A,V1 & A,70 \\ Y^{nd} & Y,0 & Y,0 & Y,7 & Y9,7 & A,01 & 17,\xi & Y7,7 & A,V1 & A,70 \\ Y^{nd} & Y,0 & Y,0 & Y,7 & Y9,7 & A,01 & 17,\xi & Y7,7 & A,V1 & A,70 \\ Y^{nd} & Y,0 & Y,0 & Y,0 & Y,0 & Y,0 & Y,0 & Y,1 & Y,1 & X,1 \\ Y^{nd} & Y,0 & Y,0 & Y,0 & Y,0 & Y,0 & Y,1 & Y,1 & X,1 \\ Y^{nd} & Y,0 & Y,0 & Y,0 & Y,0 & Y,0 & Y,1 & X,1 & X,1 \\ Y^{nd} & Y,0 & Y,0 & Y,0 & Y,0 & Y,0 & Y,1 & X,1 $	I A								,			
$T_{1} = \begin{bmatrix} r^{rd} & 7.5 & Y \cdot 1. & Y \cdot 1. & Y \cdot 1. & Y \cdot 5. & Y \cdot 7. & A.A. \\ & & & & & & & & & & & & & & & & & & $												
th o, . 15, Å £ Å, £ Å, £ Å 1., Y Y., O £ T, A Å, TY Ynd 11, £ YY, A Y., Y A, V1 1T, V YA, T Y, E A, AV Trd 1., V 13, Å T., O A, V. A, T Y., A, A Y., T A, AV A, A Y., C Y., T A, A Trd 1., V YY, T YY, T A, AV 1Y, O YY, V O, O A, A Trd V, V 1A, I TT, A A, IA A, A YY, V YY, I A, A Trd 1., A 10, I EV, E A, E Y, A 1A, A EV, T A, A Trd 1., A 10, I EV, E A, E Y, A 1A, A EV, T A, A Trd 1., A 10, I EV, E A, E Y, A 1A, A EV, T A, A Trd Y, A 17, A 10, I EV, E A, E Y, I <	-											
T1.	1 4	-						,				
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$		-										
T11 \$\frac{\pmatrix}{\pmatrix} \frac{\pmatrix}{\pmatrix} \pma	т	'							,			
$T_{11} = \begin{bmatrix} Y^{nd} & 1 \cdot , \vee & Y \gamma r & Y \gamma r & A, \vee Y & 1 \gamma , \cdot & Y \gamma , \vee & \circ, r \circ & A, \Lambda r \\ Y^{rd} & V, \vee & 1 \lambda, 1 & r \gamma, q & A, \gamma A, & A, \vee & \gamma \xi, \gamma & \gamma \xi, 1 & A, \Lambda e \\ \xi^{th} & \gamma, A & 1 \circ, 1 & \xi V, \xi & A, \xi \circ & V, A & 1 \Lambda, A & \xi V, \gamma & A, \gamma e \\ Y^{nd} & 1 \cdot, q & Y \gamma, \circ & Y \gamma, \gamma & A, V \circ & 1 r, q & Y \gamma, q & \xi, \gamma A & A, \Lambda e \\ \hline T_{17} = \begin{bmatrix} Y^{nd} & V, \cdot & 1 \lambda, r & r \gamma, \gamma & A, V \circ & q, r & \gamma \xi, \circ & Y \gamma, \circ & A, \Lambda i \\ \xi^{th} & \gamma, e & 1 \lambda, r & r \gamma, \gamma & A, V \circ & q, r & \gamma \xi, o & \gamma \gamma, e \\ Y^{nd} & 1 \circ, \circ & \gamma \circ, \xi & 1 \gamma, \circ & A, \gamma i & 1 \gamma, \circ & \gamma \gamma, 1 & \xi \gamma, \circ & A, \lambda i \\ \hline T_{17} = \begin{bmatrix} Y^{nd} & V, \circ & Y \circ, r & Y \gamma, r & A, \nabla i & 1 \gamma, \xi & Y \gamma, \gamma & A, \vee i & A, \gamma i \\ Y^{nd} & V, \circ & Y \circ, r & Y \gamma, r & A, \sigma i & 1 \gamma, \xi & Y \gamma, \gamma & A, \vee i & A, \gamma i \\ \hline \end{bmatrix}$	Ι γ.					,			,			
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$		-				,			,			
$T_{17} = \begin{bmatrix} \xi^{\text{th}} & 7, \wedge & 1 \circ, 1 & \xi \vee, \xi & \wedge, \xi \circ & \vee, \wedge & 1 \wedge, \wedge & \xi \vee, 7 & \wedge, 7 \vee \\ Y^{\text{nd}} & 1 \cdot, 9 & Y \cdot, \circ & Y \cdot, 7 & \lambda, \vee \circ & 1 \cdot, 9 & Y \cdot, 9 & \xi, \gamma \wedge & \wedge, \wedge \circ \\ Y^{\text{rd}} & \vee, \cdot & 1 \wedge, 7 & 7 \cdot, 7 & \lambda, \vee & 9, 7 & Y \cdot, \circ & Y \cdot, \circ & \lambda, \wedge 1 \vee \\ \xi^{\text{th}} & 7, \cdot & 1 \circ, \cdot & \xi \vee, \vee & \lambda, \xi 7 & \vee, Y & Y \cdot, 1 & \xi 7, \cdot & \lambda, 7, \cdot \\ Y^{\text{nd}} & 1 \circ, \circ & Y \circ, \xi & 1 1, \cdot & \lambda, 7, 1 & 17, \circ & Y \cdot, 1 & 0, \circ \vee & \lambda, \vee \circ \\ T_{17} & T^{\text{rd}} & V, \circ & Y \cdot, 7 & Y \cdot, 7 & X, V & 17, \xi & Y \cdot, Y & X, V & \lambda, 1 \circ \end{bmatrix}$	Tw								,			
$T_{17} = \begin{bmatrix} Y^{\text{nd}} & 1 & 1 & 1 & 1 & 1 & 1 & 1 & 1 & 1 &$												
$T_{17} = \begin{array}{c ccccccccccccccccccccccccccccccccccc$,						
th 1 10 £V,V A,£1 V,Y Y,1 £1, A,1 Ynd 10.0. Y0.£ 11, A,11 11,0 YV,1 0.0V A,V0 T1r Yrd V,0 Y.,r Y9,r A,01 17,£ Y1,Y A,V1 A,10	Т								,	-		
T _{1r} Y nd 10,0 Y0,£ 11,0 A,17 17,0 YY,1 0,0Y A,Y0	1 17		· ·					,	,			
T ₁₀		_										
111 17 17 17 17 17 17 17 17 17 17 17 17	T.,,	۳rd	,									
ε ^m	1 17	ź th	٦,٢		٤٧,٧	۸,٤٥	9,1	75,7	17,9	۸,٥٥		

c) Soil sodicity (ESP):

The sodification phenomenon constitutes a highly complicated problem in the studied clay soil, which constricts its productivity. Data in Table (\(\gamma\) and \(\gamma\)) showed a gradual obvious decline in ESP value with increasing the experimental time, where its value reduced below the safe limit (<\gamma\) after four periods. Data presented in Tables (\(\gamma\) and \(\gamma\)) showed that exchangeable sodium percentage (ESP) values before application amendments with soil

depth (${}^{\uparrow}{}^{\Lambda, \vee}$ %) While, after repeatedly leaching cycles and application amendments, ESP values decreased with T_{Λ} and reach about of ${}^{\uparrow}{}^{\xi}{}^{\xi}$ % and ${}^{\uparrow}{}^{\xi}{}^{\xi}$ % with relative decrease (${}^{\circ}{}^{\varepsilon}{}^{\xi}{}^{\eta}{}^{\xi}$ %) for section width of ${}^{\Lambda}{}^{\xi}{}^{\eta}{}^{\eta}{}^{\xi}{}^{\eta}{}^{\eta}{}^{\xi}{}^{\eta}{}^$

The ESP values can be arranged descending as fallow $T_1 > T_2 > T_3 > T_4 > T_4 > T_5 > T_5 > T_5 > T_6 > T_6 > T_6 > T_6 > T_7 > T_7 > T_7 > T_8 > T_8 > T_8 > T_8 > T_8 > T_9 >$

The process of leaching was effective in presence of amendments. However, ESP decreased by leaching without using amendments and the soil remained sodic with highly ESP values. However, the final ESP obtained after leaching with amendments i.e., highest R.D% in the values of ESP. The use of the T_A , T_B proved to be more effective than other treatments. Also, the removal of sodium efficiency (RSE) or percentage of Na-removed from the soil in the end of the experiment was increased after the application of the amendments, particularly T_A where RSE each to or and ir % for both the section width A,. and Nm.respectively. The RSE values can be arranged descending as fallow: $T_A > T_1 > T_2 > T_3 > T_4 > T_5 > T_6 > T_6 > T_7 > T_7 > T_7 > T_7 > T_8 > T_8 > T_8 > T_8 > T_8 > T_9 >$ T₁. Similar trend was found throughout the three seasons and in both section width of ^ m and \7 m. This could be attributed to the dominance of soluble calcium on the exchange complex which encourage decreasing of both soluble and exchangeable sodium hence decreasing the ESP values. These results are in agreement with Mansour et.al (Y·)1), Abd El-Hamid et.al $(7 \cdot 11).$

Soil physical properties

Soil physical properties are a fundamental part of soil quality assessment, as they often cannot be easily improved. Of special important, is porosity and pore size distribution. Thereby, it affects the water retention and soil hydraulic conductivity. Soil bulk density is a major product of the changes in the soil and field conditions. It is affected by the variations in soil texture, soluble salts, and exchangeable sodium percentages, all of which govern the structural status.

a) Soil hydraulic conductivity (Ks):

The distinguishing characteristics of slowly permeable saline sodic and sodic soils are due to high contents of exchangeable sodium and low hydraulic conductivity. The hydraulic conductivity measurement provides an indication of relative water transmission rate of the soil and depends on many factors, especially the volume of drainable pores. Data in Tables (^ and ^) showed that the effect. amount of soluble calcium and organic matter from previous mixtures which enhanced the soil aggregates of different treatments of the tested soil amendments under leaching cycles on hydraulic conductivity (Ks). Data reveal that the values of hydraulic conductivity (Ks), increased as a result of repeatedly leaching cycles with different

amendments. The data showed that the mean values of Ks are response to the application of different soil amendments. The data reveal that the most effective treatments were T_A , T_B and T_V for section A.m.and N m. for surface layer.

Table A: Some physical properties of soil at surface layer (· -۱º cm.) throughout the experimental time.

throughout the experimental time.												
		Sec	tion width	of ^ m.	Section width of ۱٦ m.							
Treatments	Periods	B.D	P.R	Ks	B.D	P.R	Ks					
		(kgm ^{-*})	.(kg cm ^{-*.)}	(cm hr ⁻ ')	(kgm ^{-r})	(kg cm ⁻¹ .)	(cm hr ⁻ ')					
Intermittent leaching only.	\ st	1,£9	٥٣,٤	٠,٥٠	1,0.	01,7	Ο, έλ					
	۲nd	١,٤٦	٥٠,٧	٠,٥٦	١,٤٧	01,7	.,01					
Ty Control	۳ rd	١,٤٢	٤٨,٥	٠,٦٥	١,٤٣	٤٦,٨	۲۰،۰۲					
(Leaching only)	٤ th	١,٤٠	٤٤,٤	٠,٧٢	١,٤١	٤٢,٩	٠,٦٢					
Ττ	۲na	١,٤٤	٤٨,٧	٤٢،٠	١,٤٦	٥٠,١	٠,٦٠					
	۳rd	1,89	٤٦,٣	٠,٦٦	١,٤١	٤٦,٦	• • • • • • •					
	£ th	۱٫۳۸	٤٢,٥	٠,٨٠	۱٫۳۸	٤١,٩	٠,٨٥					
Тτ	Y nd	١,٣١	٤١,٥	٠,٩٠	١,٣٩	٤٤,١	٠,٨٨					
	۳ rd	۱,۲۸	٣٩,٨	١,٢٠	100	٤٢,٢	٠,٩٦					
	£ th	١,٢٦	٣٨,٤	١,٤٠	۱٫۳۰	۳۸,٦	1,•1					
	۲nd	١,٣٣	۳۸,۱	١,٠٦	1,50	٤٠,٩	٠,٩٦					
T٤	۳ rd	۱,۲۸	٣٦,٢	1,50	١,٣٠	٣٨,٦	1,.٣					
	ź th	1,77	٣٣,٤	1,00	۲۲,۱	٣٢,٧	١,٢٠					
T.	۲ nd	1,79	٣٩,٤	٠,٩٥	۲۳,۲۱	٤١,٦	٠,٩٠					
	۳rd	1,77	٣٧,٥	1.70	١,٣٢	٣٨,٨	٠,٩٩					
	٤ th	1,70	٣٥,٩	١,٥٠	۱,۲۸	٣٥,٤	١,١٦					
	۲nd	1,50	٤٤,٤	٠,٨٢	١,٤١	٤٦,٧	٠,٧٨					
T,	۳ rd	١,٣٠	٤٣,٢	٠,٩٣	۱٫۳۸	٤٣,٦	٠,٨٥					
	£ th	1,79	٤٠,٣	1,1.	١,٣٤	٣٩,٦	٠,٩٦					
	۲nd	١,٣٤	٣٤,٢	1,70	١,٢٦	٣٤,٥	١,٢٠					
Τ _Υ	۳ra	1,70	۳۳,۸	1,08	1,77	٣١,١	1,70					
	٤ th	1,10	٣٠,١	1.77	١,٢٠	٣٠,١	۱,۳۸					
	۲ nd	١,٣٠	۳۱,٦	١,٤٨	1,77	٣١,٩	1,77					
T _A	۳rd	1,19	٣٠,٢	١,٦٠	1,19	٣٠,٠	۱,۳۸					
	£ th	1,17	79,0	١,٨٠	1,10	۲۹,۰	١,٤٦					
	۲nd	1,77	٣٢,٢	1,58	1,70	٣٢,٤	١,٢٤					
T ₃	۳ra	١,٢٠	٣٠,٩	1,00	١,٢١	٣٠,٣	١,٣٠					
	£ th	.1,17	79,7	١,٧٦	1,19	79,7	١,٤٠					
	₹ na	١,٤٤	٤٦.٨	٠,٦٠	١,٤٦	٥٠,٠	۰,٦٨					
T.,	۳ ^{ra}	١,٤٠	٤٤,٣	٠,٦٩	1,£7	٤٥,١	۰,۸۳					
	٤ th	١,٣٦	٤٢,٣	٠,٨٢	١,٣٦	٤٠,٦	٠,٩٢					
	۲ nd	1,57	٣٦,٩	1,77	١,٣٢	٣٨, ٤	١,٠٦					
Τ,,,	۳ rd	١,٢٦	٣٤,٦	1, £1	۱,۲۸	٤٦,٤	1,17					
	٤ th	١,٢٠	٣٢,٣	١,٦٠	1,70	٣٢,٠	١,٢٦					
Tıx	۲ nd	١,٢٦	٣٥,٨	1,٣٠	١,٢٨	٣٦,٣	1,14					
	۳ rd	1,71	۳۲,٦	١,٥٠	١,٢٤	٣٤,٣	1,71					
	٤ th	1,17	۳۰,٥	١,٧٠	1,71	٣١,١	1,77					
	۲nd	١,٣٠	٣٦,٠	1,70	١,٣٠	۳٧,٦	1,10					
T,4	۳rd	1,77	٣٤,٠	١,٥٠	1,۲٦	٣٤,٨	۱,۲۰					
	٤ th	1,14	۳۱,۳	1,70	1,77	71,0	1,80					
L	l .	<u> </u>			·	<u> </u>						

Ks: Hydraulic conductivly(cm hr) B.D: soil bulk density (gcm) P.R: Penetration resistance kgcm

Table 1: Some physical properties of soil at sub-surface layer (10-7)

cm.) through out the experimental time.

	,,,, till t				TIME.				
Tuestine aut -	Dania de		on width o		Section width of \\ m.				
Treatments	Periods	B.D (kgm ^{-*})	P.R (kg cm ⁻ ′.)	Ks (cm hr ⁻ ') ⁽	B.D (Mgm ^{-*})	P.R (kg cm ^{-۲} .)	Ks (cm hr ⁻ ')		
Intermittent leaching only.) st	1,01	07,1	۰,۳۸	1,08	٥٣,٧	۰,۳٥		
T ₁ Control	۲na	1,59	٥٠,٧	٠,٤٥	1,0,	01,7	٠,٤٢		
(Leaching only)	۳ra	1,£7	٤٩,٥	٠,٥٢	1,54	٥٢,٨	.,0.		
(====,,,	٤ th	1,57	٤٨,٤	٠,٥٨	1,55	٤٧,٩	•,01		
Ττ	۲ nd	1,50	٤٨,٧	1,00	1,£7	٤٨.١	٠,٥٣		
• '	۳ rd	1,77	٤٦,٣	٠,٦٣	١,٣٨	٤٨,٦	٠,٥٧		
	٤th	1,57	٤١,٥	٠,٦٩	1,50	٤٣,٩	٠,٦٠		
T۳	۲nd	١,٣٨	٤٢,٥	٠,٧٥	1,58	٤٥,٧	٠,٧١		
	۳ rd	1,79	٤٠,٨	٠,٩١	1,71	٤١,٢	٠,٨٧		
	٤ th	1,77	٣٨,٤	1,.0	1,79	٤٠,٦	٠,٩٥		
T،	۲ nd	1,71	٤٠,٢	٠,٨٥	1,50	٤٣,٩	٠,٦٧		
	۳ rd	١,٢٦	٣٧,٢	٠,٩٧	١,٣٢	٤١.٦	۰,۸۱		
	٤ th	1,70	٣٥,٤	1,10	١,٣٠	۳۸,۷	٠,٩٦		
T۰	۲ nd	١,٣٤	٤٠,٤	٠,٨٠	١,٣٦	٤٢,٦	٠,٧٦		
	۳ rd	۱,۲۸	٣٨,٠	٠,٩٥	١,٣٠	٤١,٨	٠,٩٠		
	٤th	1,77	٣٦,١	1,1.	١,٢٨	٣٨,٤	١,٠٠		
Τı	₹nd	١,٤٠	٤٤,٤	٠,٦٢	1 £ Y	٤٦,٧	٠,٦٠		
	۳ra	١,٣٢	٤٤,٢	٠,٨٠	1,70	٤٨,٦	٠,٧٤		
	٤ th	1,79	٤١,٣	٠,٨٤	1,87	٤٣,٦	٠,٧٧		
Τv	۲ nd	١,٢٦	٣٦,٢	٠,٥٨	١,٣٢	۳۸,٥	٠,٨١		
	۳ rd	1,77	۳۱,۸	1,14	١,٢٦	٣٣,٩	٠,٩٣		
	٤ th	1,14	۲٥,٧	1,70	171	۲۸,۱	1,17		
T _A	۲nd	١,٢٣	۳۲,٦	٠,٥٦	1771	۳٦,٥	٠,٥٢		
	۳ rd	١,٢٠	٣٠,٢	٠,٧٠	1,78	۳۲,٦	٠,٦٦		
	٤ th	1,17	19,0	٠,٨٧	1,18	٣٠,٢	٠,٨٠		
T۹	۲ nd	1,70	٣٤,٢	1,.0	1,77	٣٦.٤	٠,٨٦		
	۳ rd	1,71	٣٠,٩	1,11	١,٢٤	٣٣,٣	٠,٩٥		
	٤ th	١,١٦	۲۹,٦	١,٤٠	١,٢٠	٣٣,٢	1,14		
Τ,.	۲ nd	1,57	٤٦,٨	٠,٥٧	١,٤٤	٤٨,٣	٠,٤٤		
	۳rd	١,٣٤	٤٤,٧	٠,٧٨	١,٣٦	٤٩,١	٠,٥١		
	٤ th	۱,۳۱	٤١,٣	٠,٧٢	١,٣٢	٤٦,٦	۰,٦٥		
Tn	۲ nd	17.	٣٩,٩	٠,٩٢	١,٣٣	٤٣,٤	٠,٦٠		
	۳ rd	١,٢٦	٣٦,٦	1,.0	١,٣١	٣٩.٠	٠,٨٠		
	٤ th	١,٢٤	٣٣,٣	1,77	1,77	٣٦,٩	٠,٩٢		
T17	۲ nd	۱,۲۸	۳۷,۱	1,.0	١,٣٠	٣٩,٣	٠,٩٩		
	۳ rd	1,70	۳۲,٦	1,17	١,٣٢	٣٤,٠	٠,٩٨		
	٤ th	١,٢٠	۳۰,٥	١,٣٠	1,70	٣٢,١	1.0		
T,17	۲nd	1,۲9	۳۷,٥	٠,٩٦	١,٣٢	٣٩,٢	٠,٧٥		
	۳rd	1,70	٣٤,٣	1,1•	۱,۲۸	٣٦,٤	٠,٩٢		
	٤ th	1,77	۳۱,٦	1,70	١,٢٦	۳۲,۸	٠,٩٦		

Ks: Hydraulic conductive

B.D: soil bulk density

P.R: Penetration resistance

The Ks values can be arranged ascending as follows $T_1 < T_2 < T_3 < T_4 < T_4 < T_5 < T_5 < T_6 < T_6 < T_7 < T_7 < T_7 < T_7 < T_8 < T_8 < T_8 < T_9 < T$

properties and the dynamic soil water movement (EL-Sharawy *et.al.* ۲۰۰۸ and Mansour ۲۰۱۲). Further more surface applied water would pass through the surface applied amendment and infiltrate the top layers allowing exchange process between Ca^{۲+} and Na⁺ (El-Sharawy *et al.*, ۲۰۰۸).

b). Penetration Resistance (P.R.):

Penetration resistance was measured with a cone number (¹) and cross-sectional area of ¹ cm². Mean values of the penetration resistance in the surface and subsurface layers throughout three periods are presented in tables (^ and ¹). The data showed that the P.R. decrease with repeatedly leaching cycles and application of different soil amendments either surface or subsurface layers. The data reveal that the most effective treatments were T_A , T_A and T_V for section width of ^.m and ^7 m. The P.R. values can be arranged accordingly as follow:

 $T_A < T_Y < T_{YY} < T_{YY}$

c) Soil bulk density:

Soil bulk density is the main soil character that must be taken into consideration when improving soil physical properties, especially in such clayey soil. Data in Tables (^ and ^) showed that the impact of open drains installation and the secondary treatments of soil amendments (alternatives gypsum) on soil bulk density were more pronounced. The data indicated that the process of leaching was effective in presence as well as in absence of amendments. Where, it is noticed that the values of soil bulk density were reduced with time, increasing leaching cycles and applied alternatives gypsum compare with initial soil.

The obtained data showed that there was a reduction in soil bulk density after three seasons of adding alternatives gypsum as an amendment from 1,59 at the initial state to 1,11, 1,17 and 1,10 g cm at the applied mixtures of T_{Λ} , T_{3} and T_{V} , respectively, for section width Λ_{V} , while, the mean values reach to 1,1°, 1,19 and 1,1' for section width 17m, at the same applied mixtures T_V , T_A and T_1 . The bulk density improved as a result of amended soil with all the treatments. Application of different soil amendments decreased the soil bulk density compared with the (T1) leaching only and they could be arranged descending as follows: $T_1 > T_7 > T_1 > T_7 > T_7 > T_9 >$ $T_4 > T_{11} > T_{17} > T_{17} > T_{17} > T_{4} > T_{5}$. Similar trend was found throughout the three seasons and both for section width A,. and 17 m. either in surface and subsurface. These results may be attributed to the decomposition amendments and increasing exchangeable calcium which enhance aggregation process and consequently increase apparent soil bulk density volume and decrease soil bulk density which increased the efficiency of leaching processes (Abd El-Hamid et al., ۲۰۱۱).

d) Water table depth (WTD):

Water table depth fluctuations during leaching cycles are illustrated in Figs. ($^{\Upsilon}$ and $^{\Upsilon}$). The data showed that water table depth increased with decreasing spacing between drainage, and increasing the leaching cycles as well as develops more rapidly in the presence of the applied amendment treatments than the control, particularly T_{\wedge} of application. WTD fluctuated between $^{\Lambda}$ to $^{\Upsilon}$ cm and $^{\Upsilon}$ to $^{\Upsilon}$ cm for both drain spacing of $^{\Lambda}$.m and $^{\Upsilon}$ m., respectively at the end of experimental time (four periods) for application previous different amendments. These results may be attributed to the increasing exchangeable Ca^{\dagger} which encourage flocculation of soil particles leading to the formation of large soil aggregates with void volume which increased the efficiency of leaching processes. (Mansour $et al., ^{\Upsilon} \cdot ^{\Upsilon}$)

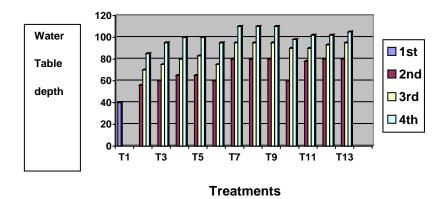


Fig. (*): Water table depth after application of different soil amendments and leaching process at drain spacing of ^ m.

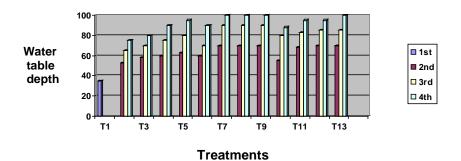


Fig. (*): Water table depth after application different of soil amendments and leaching process at drain spacing of '\' m.

CONCLUSION

From the mentioned above discussion,it could be concluded that using any amendments (.Sugar lime, vinasse, by-pass, pyrite (FeS_T) and aluminum sulfate (AI_T (SO_E)_T \^AH_TO) either individual or interaction among them tested in this study were effective in saline-sodic soil reclamation with intermittent leaching cycles .The application of any previous mixtures \(^{T-Q}\) with recommended dose, were superior to the gypsum of \(^{T}\). Mg fed \(^{T-Q}\). Generally, it could be concluded that the application any amendment precedent under intermittent leaching cycles, improved the physical properties of the soil (Ks, B.D, P.R and WTD) and chemical properties, (EC, pH and ESP).

Thus, the use of any amendment precedent may be useful for the saline-sodic soils. Therefore, the choice of any chemical amendment at any location depends on its cost, availability handling and application difficulties, relative effectiveness as judged from the improvement in soil properties and crop growth and time needed to effectively replace the adsorbed Na⁺. On the other hand, Previous mixtures and used as an external source of Ca⁺⁺, low price and easy application equal gypsum.

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كفاءة استخدام مخلفات المصانع المختلفة كمصلحات للأراضى الملحية الصودية صبحي فهمي منصور، محمد رضا أحمد محمود ، محمد محمد حسن حماد و الحسيني المرسى السيد خفاجي

معهد بحوث الأراضى والمياه والبيئة- مركز البحوث الزراعية- الجيزة- مصر

أقيمت تجربة حقلية بمحطة بحوث جنوب سهل الحسينية بمحافظة بورسعيد خلال عامين متتاليين للمناعدة وتحويلها إلى محسنات تربة كبدائل للجبس لذا تم تقييم تأثير إحدى عشر مركباً تم تحضيرها من المخلفات الآتية: (الفيناس- جير السكر-الباى باس) بالإضافة إلى الشبه (كبريتات الالومنيوم) و البيريت (كبريتيد الحديد) ومقارنة تأثير كلا منهم بالجبس و ذلك على سرعة استصلاح الأراضي الملحية الصودية مع إتباع الغسيل المتقطع بمياه ترعة السلام

وقد أشارت النتائج المتحصل عليها على:أن هناك تحسن واضح نتيجة إضافة كل المحسنات السابقة على الخواص الطبيعية (الخواص الكيميائية للتربة (ملوحة التربة – قلوية التربة - نسبة الصوديوم المتبادل) والخواص الطبيعية (الكثافة الظاهرية – اندماج التربة – منسوب الماء الأرضي) حيث تقوقت المعاملات أرقام $\sim 10^{10}$ و $\sim 10^{10}$ الترتيب عن باقي المعاملات سواء في الطبقة السطحية أو التحت السطحية ويمكن ترتيب تأثير هذه المعاملات $\sim 10^{10}$ المعاملات $\sim 10^{10}$

 $T_1 < T_7 < T_1 < T_7 < T_7$

 $(Y), \qquad (Y), \qquad (Y),$

(4), (4), (4), (4), (4), (4), (4),